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carbonate: calcium carbonate  $\rightarrow$   
calcium oxide + carbon dioxide

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sulphate When you put a little  
amount of blue copper sulphate

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in a clean test tube, then heat the content of the test by using a flame, A black substance will be formed, Blue copper sulphate decomposes by the heat into copper oxide (black colour) and sulphur trioxide.  $\text{CuSO}_4 \cdot \text{H}_2\text{O} \rightarrow \text{CuO} + \text{SO}_3 + \text{H}_2\text{O}$

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