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aqueous solution: a solution in which the solvent is water: solvent: the dissolving medium in a solution: surfactant: wetting agent that interferes with hydrogen bonding in water: strong electrolyte: a substance that completely dissociates into its ions in solution: water of hydration: the water loosely held in a crystal structure:

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NEW YORK CITY TRANSIT AUTHORITY HAZARDOUS WASTE WORKBOOK

Question 3 - Impacts on Surface Water -
Full EAF (Part 2) Full Environmental

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Assessment Form (FEAF) Workbook. The proposed action may affect one or more wetlands or other surface waterbodies (e.g., streams, rivers, ponds or lakes).

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the DfE), was led by Dr Wayne Tennent of the University of East London.

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materials, geochemistry, and environmental applications. The authors' goal is to present the material at a level that serves both the graduate student seeking to learn the state of the art, and also the industrial engineer or chemist seeking to develop additional expertise or to find the data needed to solve a specific

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problem. The wide range of people for whom this topic is important provides a challenge. Advanced work in this area is distributed among physical chemists, chemical engineers, geochemists, and other specialists, who may not be aware of parallel work by those outside their own specialty. The particular aspects of high-

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temperature aqueous physical chemistry of interest to one industry may be irrelevant to another; yet another industry might need the same basic information but in a very different form. To serve all these constituencies, the book includes several chapters that cover the foundational thermophysical properties (such as gas

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solubility, phase behavior, thermodynamic properties of solutes, and transport properties) that are of interest across numerous applications. The presentation of these topics is intended to be accessible to readers from a variety of backgrounds. Other chapters address fundamental areas of more specialized interest, such as

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critical phenomena and molecular-level solution structure. Several chapters are more application-oriented, addressing areas such as power-cycle chemistry and hydrothermal synthesis. As befits the variety of interests addressed, some chapters provide more theoretical guidance while others, such as those on

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acid/base equilibria and the solubilities of metal oxides and hydroxides, emphasize experimental techniques and data analysis.

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and interactions among small numbers of water molecules. It is mostly the result of the author's own research spanning over 40 years in the field of aqueous solutions. An understanding of the properties of liquid water is a prelude to the understanding of the role of water in biological systems and for the evolution

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of life. The book is targeted at anyone who is interested in the outstanding properties of water and its role in biological systems. It is addressed to both students and researchers in chemistry, physics and biology.

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The Radiation Chemistry of Water tackles radiation-induced changes in water and explains the behavior of irradiated water, with some changes in aqueous solutions.

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This book deals primarily with short-lived species like the hydroxyl radical, hydrated electron, and hydrogen atom, which cause the chemical changes in irradiated water and aqueous solutions. These species and their origin, properties, and dependence of their yields on various factors are discussed in several chapters. Other topics

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also covered are the diffusion-kinetic model of water radiolysis and some general cases, radiation sources, and dosimetry. This book is most useful to students in the fields of radiation chemistry, physical chemistry, radiobiology, and nuclear technology.

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Continuing a trend of covering an increasingly wide range of topics associated with water, steam, and high-temperature aqueous systems, the papers in this book cover metastable states and nucleation, supercooled, superheated and stretched water, molecular modeling of aqueous systems, frontiers of physical

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chemistry of aqueous solutions, high-temperature aqueous systems including measurement techniques, hydrothermal oxidation, chemical processes in steam cycles, and plant cycle chemistry.

"The aim of this book is to explain the unusual properties of both pure liquid

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water and simple aqueous solutions, in terms of the properties of single molecules and interactions among small numbers of water molecules. It is mostly the result of the author's own research spanning over 40 years in the field of aqueous solutions."--Jacket.

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Water dominates the surface of Earth and is vital to life on our planet. It is a remarkable liquid which shows anomalous behaviour. In this Very Short Introduction John Finney introduces the science of water, and explores how the structure of water molecules gives rise to its physical and chemical properties. Considering

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Water in all three of its states as ice and steam as well as liquid, Finney explains the great importance of an understanding of its structure and behaviour to a range of fields including chemistry, astrophysics, and earth and environmental sciences.

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